

## Estimate of (n, p) Cross-section for Unstable Nuclide $^{53}\text{Mn}$

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### Introduction

$^{53}\text{Mn}$  is one of the long-lived radionuclides (decay by electron capture) having half life of  $3.74 \times 10^6$  year, produced inside the fusion reactor, due to transmutation of stable isotopes of Stainless Steel (SS) present in the structural materials [1,2]. Due to its longer half life and as primary nuclide (isotopes having percentage contribution  $\geq 50\%$ ) it interacts with neutrons inside the reactor and give rise to different nuclear reactions depending on the neutron energy and reaction Q values. Neutrons emitted in a D-T fusion reactor ( $\text{D}+\text{T} \rightarrow \text{n}+\alpha+17.6 \text{ MeV}$ ) are of 14.1 MeV energy, however inside the fusion reactor the energy of the neutrons degrades due to interactions with various reactor materials resulting in a neutron spectrum with energy from eV to MeV range. From neutronics point of view, it is therefore important to study the nuclear reactions such as  $^{53}\text{Mn}(\text{n,p})$ ,  $^{53}\text{Mn}(\text{n},\alpha)$ ,  $^{53}\text{Mn}(\text{n,d})$  and  $^{53}\text{Mn}(\text{n,t})$ . The (n,p) (n,  $\alpha$ ) and (n,d) reactions can cause the production of hydrogen, helium and deuterium gases while (n,t) is important for production of tritium ( $^3\text{H}$ ;  $T_{1/2}=12.33$  year) inside the fusion reactor [3]. The (n,p) and (n, $\alpha$ ) reactions are more critical because He/H deposit at different locations inside the reactor can degrade the integrity of the materials.

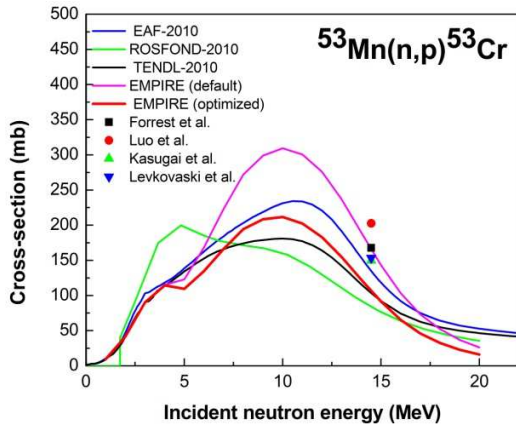
$^{55}\text{Mn}$  is the only one stable isotope found in nature with abundance 100%.  $^{53}\text{Mn}$  is radioactive and is produced during the operation of the fusion reactor through different pathways given by  $^{54}\text{Fe}(\text{n,np})$ ,  $^{54}\text{Fe}(\text{n,2n})^{53}\text{Fe}(\beta^+)$ ,  $^{54}\text{Fe}(\text{n,d})$  [4]. Neutron induced cross-section measurements by direct neutron activation method is not feasible because  $^{53}\text{Mn}$  target cannot be made and

therefore, one needs indirect methods to estimate these cross sections.

### Experimental Details

Recently measurement of  $^{55}\text{Fe}(\text{n,p})$  reaction cross-section has been carried out using surrogate technique. In this method, the transfer reaction  $^{52}\text{Cr}(\text{}^6\text{Li,d})^{56}\text{Fe}^*$  is measured which is the surrogate of the desired reaction  $^{55}\text{Fe}(\text{n,p})$  [5] leading to same compound system  $^{56}\text{Fe}^*$ . In the experiment, the deuteron (PLF) was detected in coincidence with protons coming as evaporation from  $^{56}\text{Fe}^*$ . It is observed that in the same experiment, another PLF  $\alpha$ -channel is also present given by transfer reaction  $^{52}\text{Cr}(\text{}^6\text{Li},\alpha)^{54}\text{Mn}^*$ . This is the surrogate of the  $^{53}\text{Mn}(\text{n,p})$  reaction. In this paper, this surrogate reaction has been studied to measure  $^{53}\text{Mn}(\text{n,p})$  reaction cross-section by using nuclear reaction modular codes EMPIRE and TALYS [6] and compare the values with systematics [7], and evaluated data libraries [8]. Table-1 shows the calculation for all isotopes of Fe and Mn. Comparison of the calculated cross-sections with available evaluated data libraries and systematics has been shown in Fig.1 for  $^{53}\text{Mn}(\text{n,p})$ . The experimental data is under analysis to determine the  $^{53}\text{Mn}(\text{n,p})$  reaction cross-section data by measuring the alpha (PLF) in singles and alpha and proton (evaporation from  $^{54}\text{Mn}^*$ ) in coincidence. Transfer reaction  $^{45}\text{Sc}(\text{}^6\text{Li},\alpha)^{47}\text{Ti}^*$  has been chosen as surrogate of the known (for which experimental data is available)  $^{46}\text{Ti}(\text{n,p})$  reaction, which is the reference reaction for both the above desired reaction ( i.e.  $^{55}\text{Fe}(\text{n,p})$  and  $^{53}\text{Mn}(\text{n,p})$  ) to be used in the surrogate ratio method. All the detailed information about three

reactions,  ${}^6\text{Li}+{}^{52}\text{Cr}\rightarrow\text{d}+{}^{56}\text{Fe}^*$  [surrogate of  ${}^{55}\text{Fe}(\text{n,p})$ ],  ${}^6\text{Li}+{}^{52}\text{Cr}\rightarrow\alpha+{}^{54}\text{Mn}^*$  [surrogate of  ${}^{53}\text{Mn}(\text{n,p})$ ] and  ${}^6\text{Li}+{}^{45}\text{Sc}\rightarrow\alpha+{}^{47}\text{Ti}^*$  [surrogate of  ${}^{46}\text{Ti}(\text{n,p})$ , also known as reference reaction in the present experiment] will be presented.



**Fig.1** (color online) Comparison of calculated excitation function of  ${}^{53}\text{Mn}(\text{n,p})$  with systematics and evaluated data files.

**References**

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**TABLE 1-** Comparison of Nuclear model calculations and semi-empirical formula (Forrest, 1986) for **Fe** and **Mn** cross-section at 14.5 MeV

Reaction	Model Calculations (mb)		Semi-emp.[7] formul a (S) (mb)	Exp. data [11] (E) (mb)	Deviation		
	EMPIRE (C1)	TALYS (C2)			(C1- E) /E	(C2- E) /E	(S- E) /E
${}^{54}\text{Fe}(\text{n,p})$	301	295	355	$315\pm 10$	-0.044	-0.063	0.126
${}^{55}\text{Fe}(\text{n,p})$ our experiment	217	198	190	$189\pm 15$ our measurement	0.148	.047	.0052
${}^{56}\text{Fe}(\text{n,p})$	106	112	101	$108\pm 3$	-0.019	0.037	-0.065
${}^{57}\text{Fe}(\text{n,p})$	58	61	54	$80\pm 10$	-0.275	0.238	0.325
${}^{53}\text{Mn}(\text{n,p})$	106.57	103.31	167.79	Exp.data analysis in progress			
${}^{55}\text{Mn}(\text{n,p})$	34.79	26.91	45.30	63	-0.447	-0.573	-0.280